

# Ethyl 1-benzenesulfonyl-2-[(*E*)-2-(2-methylphenyl)ethenyl]indole-3-carboxylate

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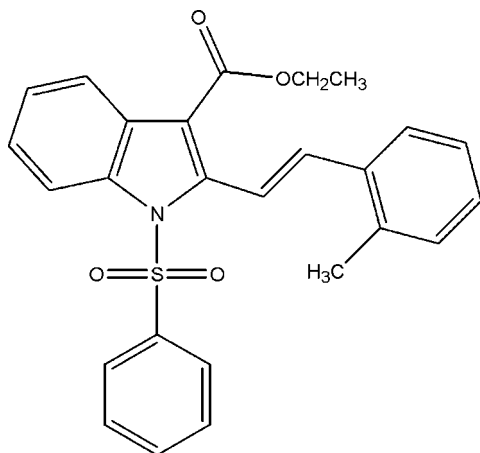
Received 2 January 2011; accepted 12 January 2011

Key indicators: single-crystal X-ray study; *T* = 295 K; mean  $\sigma(\text{C}-\text{C})$  = 0.003 Å; *R* factor = 0.042; *wR* factor = 0.117; data-to-parameter ratio = 16.7.

In the title compound, C<sub>26</sub>H<sub>23</sub>NO<sub>4</sub>S, the phenyl, tolyl and ester groups make dihedral angles of 82.28 (5), 77.67 (6) and 8.52 (6)°, respectively, with the indole ring system. The S atom of the sulfonyl group is displaced by 0.1968 (4) Å from the indole mean plane. The molecular structure is stabilized by weak intramolecular C—H···O interactions. The crystal structure features short intramolecular C—H···O contacts and  $\pi$ - $\pi$  stacking interactions between the phenyl and tolyl groups [centroid-centroid distance = 3.9448 (11) Å].

## Related literature

For the biological activity of indole derivatives, see: Andreani *et al.* (2001); Kolocouris *et al.* (1994); Merck (1973). For the structures of closely related compounds, see: Chakkaravarthi *et al.* (2007, 2008).



## Experimental

### Crystal data

C<sub>26</sub>H<sub>23</sub>NO<sub>4</sub>S  
*M<sub>r</sub>* = 445.51  
 Monoclinic, *P*2<sub>1</sub>/*n*  
*a* = 10.4248 (4) Å  
*b* = 8.3629 (3) Å  
*c* = 25.2284 (11) Å  
 $\beta$  = 92.902 (1)°  
*V* = 2196.63 (15) Å<sup>3</sup>  
*Z* = 4  
 Mo *K*α radiation  
 $\mu$  = 0.18 mm<sup>-1</sup>  
*T* = 295 K  
 0.24 × 0.20 × 0.18 mm

### Data collection

Bruker Kappa APEXII diffractometer  
 Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996)  
*T<sub>min</sub>* = 0.958, *T<sub>max</sub>* = 0.968  
 22952 measured reflections  
 4848 independent reflections  
 3551 reflections with *I* > 2σ(*I*)  
*R<sub>int</sub>* = 0.029

### Refinement

$R[F^2 > 2\sigma(F^2)]$  = 0.042  
 $wR(F^2)$  = 0.117  
*S* = 1.03  
 4848 reflections  
 291 parameters  
 H-atom parameters constrained  
 $\Delta\rho_{\text{max}}$  = 0.25 e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}}$  = -0.28 e Å<sup>-3</sup>

Table 1

Hydrogen-bond geometry (Å, °).

<i>D</i> —H··· <i>A</i>	<i>D</i> —H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> —H··· <i>A</i>
C10—H10···O3	0.93	2.48	3.000 (3)	116
C13—H13···O1	0.93	2.33	2.908 (3)	120
C16—H16A···O1 <sup>1</sup>	0.97	2.56	3.381 (3)	143

Symmetry code: (i) *x* - 1, *y*, *z*.

Data collection: *APEX2* (Bruker, 2004); cell refinement: *SAINT* (Bruker, 2004); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *PLATON* (Spek, 2009); software used to prepare material for publication: *SHELXL97*.

CR wishes to acknowledge AMET University management, India, for their kind support.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: GK2338).

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**supplementary materials**

*Acta Cryst.* (2011). E67, o448 [ doi:10.1107/S1600536811001863 ]

## Ethyl 1-benzenesulfonyl-2-[(*E*)-2-(2-methylphenyl)ethenyl]indole-3-carboxylate

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### Comment

Indole derivatives exhibit antihypertensive (Merck, 1973), antitumour (Andreani *et al.*, 2001) and antiviral (Kolocouris *et al.*, 1994) activities. The geometric parameters of the title molecule (Fig. 1) agree well with the reported similar structures (Chakkaravarthi *et al.* 2007, 2008).

The phenyl ring makes the dihedral angle of 82.28 (5)° with the indole ring system. The benzene ring (C20—C25) forms the dihedral angle of 77.67 (6)° with the indole ring system. The S atom of the sulfonyl group is displaced 0.1968 (4) Å from the indole mean plane. The sum of the bond angles around N1 [358.5 (1)°] indicates that N1 atom is  $sp^2$  hybridized. The molecular structure is stabilized by weak intramolecular C—H···O interactions and the crystal packing exhibits weak intermolecular C—H···O (Fig.2 and Table 1) and  $\pi$ – $\pi$  interactions [ $Cg2 \cdots Cg4$  ( $1/2 - x, -1/2 + y, 1/2 - z$ ) distance of 3.9448 (11) Å;  $Cg2$  and  $Cg4$  are the centroids of C1—C6 ring and C20—C26 ring, respectively] .

### Experimental

To a suspension of hexane (5 ml) washed NaH (0.29 g, 6.10 mmol) in dry THF (10 ml) at -10° C under N<sub>2</sub> atmosphere was slowly added the solution of diethyl (3-(ethoxycarbonyl)-1-phenylsulfonyl-1*H*-indol-2-yl)methylphosphonate (0.97 g, 2.03 mmol) in dry THF (5 ml) *via* syringe and stirred for 15 min. Then a solution of 2-methylbenzaldehyde (0.28 g, 2.32 mmol) in dry THF (5 ml) was added and allowed to stir for additional 2 h. After completion of the product formation (monitored by TLC), it was then poured over crushed ice (100 g) containing conc. HCl (3 ml). The solid formed was filtered and recrystallized with MeOH to afford ethyl 2-(2-methylstyryl)-1-phenylsulfonyl -1*H*-indole-3-carboxylate as bright yellow crystals [0.70 g, 78%; melting point 371–373 K].

### Refinement

H atoms were positioned geometrically and refined using riding model approximation with C—H = 0.93 Å and  $U_{iso}(H) = 1.2U_{eq}(C)$  for aromatic C—H and C—H = 0.97 Å and  $U_{iso}(H) = 1.2U_{eq}(C)$  for methylene group and C—H = 0.96 Å and  $U_{iso}(H) = 1.5U_{eq}(C)$  for methyl group.

## Figures

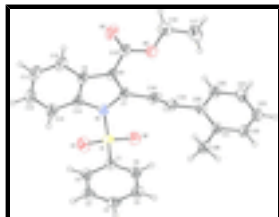


Fig. 1. The molecular structure of the title compound with atom labels and 30% probability displacement ellipsoids for non-H atoms.

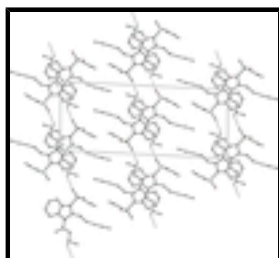


Fig. 2. The crystal packing viewed down the *b* axis. Intermolecular C-H...O interactions (hydrogen bonds) are shown as dashed lines. H atoms not involved in hydrogen bonding have been omitted.

## Ethyl 1-benzenesulfonyl-2-[(*E*)-2-(2-methylphenyl)ethenyl]indole-3-carboxylate

### Crystal data

$C_{26}H_{23}NO_4S$

$M_r = 445.51$

Monoclinic,  $P2_1/n$

Hall symbol: -P 2yn

$a = 10.4248$  (4) Å

$b = 8.3629$  (3) Å

$c = 25.2284$  (11) Å

$\beta = 92.902$  (1)°

$V = 2196.63$  (15) Å<sup>3</sup>

$Z = 4$

$F(000) = 936$

$D_x = 1.347$  Mg m<sup>-3</sup>

Mo  $K\alpha$  radiation,  $\lambda = 0.71073$  Å

Cell parameters from 31039 reflections

$\theta = 2.1$ – $31.4$ °

$\mu = 0.18$  mm<sup>-1</sup>

$T = 295$  K

Block, yellow

$0.24 \times 0.20 \times 0.18$  mm

### Data collection

Bruker Kappa APEXII  
diffractometer

Radiation source: fine-focus sealed tube  
graphite

$\omega$  and  $\phi$  scans

Absorption correction: multi-scan  
(*SADABS*; Sheldrick, 1996)

$T_{\min} = 0.958$ ,  $T_{\max} = 0.968$

22952 measured reflections

4848 independent reflections

3551 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.029$

$\theta_{\text{max}} = 27.1$ °,  $\theta_{\text{min}} = 2.1$ °

$h = -12 \rightarrow 13$

$k = -10 \rightarrow 10$

$l = -32 \rightarrow 32$

### Refinement

Refinement on  $F^2$

Primary atom site location: structure-invariant direct  
methods

Least-squares matrix: full

$$R[F^2 > 2\sigma(F^2)] = 0.042$$

$$wR(F^2) = 0.117$$

$$S = 1.03$$

4848 reflections

291 parameters

0 restraints

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.0499P)^2 + 0.6907P]$$

$$\text{where } P = (F_o^2 + 2F_c^2)/3$$

$$(\Delta/\sigma)_{\max} < 0.001$$

$$\Delta\rho_{\max} = 0.25 \text{ e } \text{\AA}^{-3}$$

$$\Delta\rho_{\min} = -0.28 \text{ e } \text{\AA}^{-3}$$

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
C1	0.29596 (16)	0.5477 (2)	0.13482 (7)	0.0434 (4)
C2	0.33149 (19)	0.4055 (2)	0.11194 (7)	0.0516 (4)
H2	0.3705	0.4051	0.0796	0.062*
C3	0.3082 (2)	0.2640 (2)	0.13788 (8)	0.0563 (5)
H3	0.3314	0.1672	0.1229	0.068*
C4	0.2511 (2)	0.2656 (2)	0.18551 (8)	0.0582 (5)
H4	0.2353	0.1697	0.2027	0.070*
C5	0.2170 (2)	0.4074 (3)	0.20820 (8)	0.0641 (6)
H5	0.1793	0.4073	0.2408	0.077*
C6	0.23856 (19)	0.5502 (2)	0.18277 (8)	0.0549 (5)
H6	0.2147	0.6466	0.1978	0.066*
C7	0.07971 (18)	0.81521 (19)	0.06872 (6)	0.0439 (4)
C8	-0.00146 (19)	0.7883 (2)	0.02544 (6)	0.0466 (4)
C9	0.0676 (2)	0.7013 (2)	-0.01332 (7)	0.0505 (5)
C10	0.0332 (2)	0.6438 (2)	-0.06437 (7)	0.0644 (6)
H10	-0.0487	0.6611	-0.0796	0.077*
C11	0.1231 (3)	0.5617 (3)	-0.09121 (9)	0.0782 (8)
H11	0.1019	0.5246	-0.1253	0.094*
C12	0.2444 (3)	0.5327 (3)	-0.06897 (9)	0.0803 (8)
H12	0.3020	0.4733	-0.0879	0.096*
C13	0.2824 (3)	0.5898 (3)	-0.01920 (8)	0.0698 (6)
H13	0.3647	0.5719	-0.0045	0.084*
C14	0.1920 (2)	0.6752 (2)	0.00782 (7)	0.0530 (5)
C15	-0.1351 (2)	0.8425 (2)	0.01525 (7)	0.0520 (5)
C16	-0.3192 (2)	0.9600 (3)	0.04901 (9)	0.0683 (6)
H16A	-0.3749	0.8700	0.0401	0.082*
H16B	-0.3273	1.0375	0.0204	0.082*
C17	-0.3556 (3)	1.0338 (3)	0.09964 (11)	0.0900 (8)
H17A	-0.2996	1.1223	0.1081	0.135*
H17B	-0.3479	0.9558	0.1275	0.135*
H17C	-0.4427	1.0710	0.0960	0.135*
C18	0.05757 (17)	0.9015 (2)	0.11824 (6)	0.0458 (4)
H18	0.1021	0.9965	0.1247	0.055*
C19	-0.02112 (18)	0.8527 (2)	0.15390 (6)	0.0479 (4)
H19	-0.0617	0.7549	0.1479	0.057*

## supplementary materials

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C20	-0.05081 (16)	0.9393 (2)	0.20259 (7)	0.0445 (4)
C21	-0.07577 (16)	0.8570 (2)	0.24966 (6)	0.0460 (4)
C22	-0.10097 (19)	0.9451 (3)	0.29430 (7)	0.0570 (5)
H22	-0.1176	0.8917	0.3255	0.068*
C23	-0.1022 (2)	1.1089 (3)	0.29386 (8)	0.0644 (6)
H23	-0.1181	1.1652	0.3246	0.077*
C24	-0.0800 (2)	1.1895 (3)	0.24804 (9)	0.0644 (6)
H24	-0.0816	1.3007	0.2475	0.077*
C25	-0.05529 (19)	1.1053 (2)	0.20262 (8)	0.0543 (5)
H25	-0.0414	1.1607	0.1715	0.065*
C26	-0.0706 (2)	0.6777 (2)	0.25305 (8)	0.0594 (5)
H26A	-0.0929	0.6442	0.2878	0.089*
H26B	0.0146	0.6417	0.2465	0.089*
H26C	-0.1304	0.6326	0.2270	0.089*
N1	0.20131 (15)	0.75147 (17)	0.05839 (6)	0.0494 (4)
O1	0.43527 (14)	0.70935 (19)	0.07195 (6)	0.0717 (4)
O2	0.31983 (13)	0.85612 (15)	0.13962 (5)	0.0603 (4)
O3	-0.19140 (16)	0.8300 (2)	-0.02733 (6)	0.0866 (5)
O4	-0.18768 (13)	0.90745 (18)	0.05650 (5)	0.0618 (4)
S1	0.32570 (5)	0.72867 (5)	0.10266 (2)	0.05081 (15)

### Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
C1	0.0433 (9)	0.0401 (8)	0.0464 (9)	-0.0039 (7)	-0.0009 (7)	0.0010 (7)
C2	0.0587 (12)	0.0467 (10)	0.0492 (10)	0.0004 (8)	0.0014 (9)	-0.0044 (8)
C3	0.0654 (13)	0.0397 (9)	0.0624 (12)	0.0005 (9)	-0.0099 (10)	-0.0035 (8)
C4	0.0644 (13)	0.0464 (10)	0.0625 (12)	-0.0059 (9)	-0.0087 (10)	0.0116 (9)
C5	0.0771 (15)	0.0591 (12)	0.0571 (12)	-0.0003 (11)	0.0132 (10)	0.0104 (10)
C6	0.0661 (12)	0.0454 (10)	0.0538 (11)	0.0034 (9)	0.0104 (9)	0.0007 (8)
C7	0.0591 (11)	0.0362 (8)	0.0371 (8)	-0.0067 (8)	0.0108 (8)	0.0032 (7)
C8	0.0653 (12)	0.0390 (9)	0.0363 (9)	-0.0110 (8)	0.0094 (8)	0.0019 (7)
C9	0.0780 (14)	0.0366 (9)	0.0380 (9)	-0.0125 (9)	0.0134 (9)	0.0027 (7)
C10	0.1043 (17)	0.0483 (11)	0.0412 (10)	-0.0157 (11)	0.0112 (10)	-0.0031 (9)
C11	0.146 (3)	0.0495 (12)	0.0414 (11)	-0.0110 (14)	0.0231 (14)	-0.0058 (9)
C12	0.139 (2)	0.0523 (12)	0.0534 (13)	0.0117 (14)	0.0431 (15)	0.0011 (10)
C13	0.0991 (17)	0.0592 (12)	0.0537 (12)	0.0121 (12)	0.0293 (12)	0.0066 (10)
C14	0.0827 (14)	0.0383 (9)	0.0396 (9)	-0.0030 (9)	0.0199 (9)	0.0045 (7)
C15	0.0654 (12)	0.0482 (10)	0.0426 (10)	-0.0170 (9)	0.0037 (9)	0.0007 (8)
C16	0.0565 (13)	0.0702 (14)	0.0784 (15)	-0.0075 (11)	0.0065 (11)	0.0081 (12)
C17	0.0768 (17)	0.0912 (18)	0.104 (2)	0.0034 (14)	0.0249 (15)	-0.0120 (16)
C18	0.0572 (11)	0.0414 (9)	0.0389 (9)	-0.0050 (8)	0.0039 (8)	-0.0034 (7)
C19	0.0635 (12)	0.0429 (9)	0.0376 (9)	-0.0074 (8)	0.0051 (8)	-0.0013 (7)
C20	0.0440 (10)	0.0504 (10)	0.0390 (9)	-0.0034 (8)	0.0024 (7)	-0.0036 (7)
C21	0.0398 (9)	0.0587 (11)	0.0394 (9)	-0.0019 (8)	0.0019 (7)	-0.0001 (8)
C22	0.0562 (12)	0.0771 (14)	0.0380 (9)	0.0023 (10)	0.0074 (8)	-0.0026 (9)
C23	0.0654 (13)	0.0780 (15)	0.0504 (11)	0.0045 (11)	0.0079 (10)	-0.0189 (11)
C24	0.0672 (14)	0.0521 (11)	0.0745 (14)	0.0019 (10)	0.0106 (11)	-0.0151 (10)

C25	0.0615 (12)	0.0514 (10)	0.0507 (10)	-0.0026 (9)	0.0085 (9)	0.0002 (9)
C26	0.0654 (13)	0.0607 (12)	0.0528 (11)	-0.0022 (10)	0.0115 (10)	0.0078 (9)
N1	0.0647 (10)	0.0456 (8)	0.0386 (8)	0.0016 (7)	0.0102 (7)	0.0027 (6)
O1	0.0589 (9)	0.0704 (10)	0.0881 (11)	-0.0103 (7)	0.0267 (8)	0.0072 (8)
O2	0.0694 (9)	0.0408 (7)	0.0699 (9)	-0.0097 (6)	-0.0036 (7)	-0.0054 (6)
O3	0.0824 (11)	0.1226 (15)	0.0535 (9)	-0.0018 (10)	-0.0107 (8)	-0.0151 (9)
O4	0.0587 (9)	0.0764 (9)	0.0504 (8)	0.0003 (7)	0.0037 (6)	-0.0047 (7)
S1	0.0530 (3)	0.0423 (2)	0.0579 (3)	-0.0081 (2)	0.0101 (2)	0.0021 (2)

*Geometric parameters (Å, °)*

C1—C6	1.377 (3)	C15—O4	1.318 (2)
C1—C2	1.381 (2)	C16—O4	1.444 (2)
C1—S1	1.7523 (17)	C16—C17	1.485 (3)
C2—C3	1.379 (3)	C16—H16A	0.9700
C2—H2	0.9300	C16—H16B	0.9700
C3—C4	1.368 (3)	C17—H17A	0.9600
C3—H3	0.9300	C17—H17B	0.9600
C4—C5	1.371 (3)	C17—H17C	0.9600
C4—H4	0.9300	C18—C19	1.313 (2)
C5—C6	1.379 (3)	C18—H18	0.9300
C5—H5	0.9300	C19—C20	1.472 (2)
C6—H6	0.9300	C19—H19	0.9300
C7—C8	1.366 (2)	C20—C25	1.389 (3)
C7—N1	1.412 (2)	C20—C21	1.408 (2)
C7—C18	1.471 (2)	C21—C22	1.382 (3)
C8—C9	1.440 (2)	C21—C26	1.503 (3)
C8—C15	1.475 (3)	C22—C23	1.370 (3)
C9—C14	1.394 (3)	C22—H22	0.9300
C9—C10	1.404 (3)	C23—C24	1.368 (3)
C10—C11	1.368 (3)	C23—H23	0.9300
C10—H10	0.9300	C24—C25	1.380 (3)
C11—C12	1.379 (4)	C24—H24	0.9300
C11—H11	0.9300	C25—H25	0.9300
C12—C13	1.383 (3)	C26—H26A	0.9600
C12—H12	0.9300	C26—H26B	0.9600
C13—C14	1.388 (3)	C26—H26C	0.9600
C13—H13	0.9300	N1—S1	1.6789 (17)
C14—N1	1.425 (2)	O1—S1	1.4217 (14)
C15—O3	1.202 (2)	O2—S1	1.4194 (14)
C6—C1—C2	121.21 (17)	C17—C16—H16B	110.3
C6—C1—S1	119.25 (13)	H16A—C16—H16B	108.5
C2—C1—S1	119.54 (14)	C16—C17—H17A	109.5
C3—C2—C1	118.89 (18)	C16—C17—H17B	109.5
C3—C2—H2	120.6	H17A—C17—H17B	109.5
C1—C2—H2	120.6	C16—C17—H17C	109.5
C4—C3—C2	120.21 (18)	H17A—C17—H17C	109.5
C4—C3—H3	119.9	H17B—C17—H17C	109.5
C2—C3—H3	119.9	C19—C18—C7	124.04 (16)

## supplementary materials

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C3—C4—C5	120.57 (18)	C19—C18—H18	118.0
C3—C4—H4	119.7	C7—C18—H18	118.0
C5—C4—H4	119.7	C18—C19—C20	125.73 (17)
C4—C5—C6	120.19 (19)	C18—C19—H19	117.1
C4—C5—H5	119.9	C20—C19—H19	117.1
C6—C5—H5	119.9	C25—C20—C21	118.73 (16)
C1—C6—C5	118.93 (18)	C25—C20—C19	120.03 (16)
C1—C6—H6	120.5	C21—C20—C19	121.24 (16)
C5—C6—H6	120.5	C22—C21—C20	118.50 (18)
C8—C7—N1	108.42 (15)	C22—C21—C26	119.57 (17)
C8—C7—C18	129.95 (17)	C20—C21—C26	121.89 (16)
N1—C7—C18	121.54 (16)	C23—C22—C21	121.92 (19)
C7—C8—C9	108.39 (18)	C23—C22—H22	119.0
C7—C8—C15	129.07 (16)	C21—C22—H22	119.0
C9—C8—C15	122.45 (17)	C24—C23—C22	119.83 (19)
C14—C9—C10	119.08 (19)	C24—C23—H23	120.1
C14—C9—C8	107.85 (16)	C22—C23—H23	120.1
C10—C9—C8	133.1 (2)	C23—C24—C25	119.8 (2)
C11—C10—C9	118.5 (2)	C23—C24—H24	120.1
C11—C10—H10	120.8	C25—C24—H24	120.1
C9—C10—H10	120.8	C24—C25—C20	121.19 (19)
C10—C11—C12	121.6 (2)	C24—C25—H25	119.4
C10—C11—H11	119.2	C20—C25—H25	119.4
C12—C11—H11	119.2	C21—C26—H26A	109.5
C11—C12—C13	121.6 (2)	C21—C26—H26B	109.5
C11—C12—H12	119.2	H26A—C26—H26B	109.5
C13—C12—H12	119.2	C21—C26—H26C	109.5
C12—C13—C14	116.9 (2)	H26A—C26—H26C	109.5
C12—C13—H13	121.5	H26B—C26—H26C	109.5
C14—C13—H13	121.5	C7—N1—C14	108.25 (16)
C13—C14—C9	122.30 (19)	C7—N1—S1	126.20 (12)
C13—C14—N1	130.7 (2)	C14—N1—S1	124.06 (14)
C9—C14—N1	106.99 (16)	C15—O4—C16	116.92 (16)
O3—C15—O4	122.5 (2)	O2—S1—O1	120.39 (9)
O3—C15—C8	122.88 (19)	O2—S1—N1	107.11 (8)
O4—C15—C8	114.58 (16)	O1—S1—N1	105.37 (9)
O4—C16—C17	107.22 (19)	O2—S1—C1	109.27 (8)
O4—C16—H16A	110.3	O1—S1—C1	108.67 (9)
C17—C16—H16A	110.3	N1—S1—C1	104.91 (8)
O4—C16—H16B	110.3		
C6—C1—C2—C3	0.2 (3)	C18—C19—C20—C21	-146.01 (19)
S1—C1—C2—C3	179.60 (15)	C25—C20—C21—C22	-1.4 (3)
C1—C2—C3—C4	-0.2 (3)	C19—C20—C21—C22	178.96 (17)
C2—C3—C4—C5	-0.4 (3)	C25—C20—C21—C26	-178.99 (18)
C3—C4—C5—C6	0.8 (3)	C19—C20—C21—C26	1.4 (3)
C2—C1—C6—C5	0.3 (3)	C20—C21—C22—C23	0.0 (3)
S1—C1—C6—C5	-179.13 (16)	C26—C21—C22—C23	177.64 (19)
C4—C5—C6—C1	-0.8 (3)	C21—C22—C23—C24	1.0 (3)
N1—C7—C8—C9	1.62 (18)	C22—C23—C24—C25	-0.6 (3)



C18—C7—C8—C9	178.34 (16)	C23—C24—C25—C20	-0.8 (3)
N1—C7—C8—C15	-174.87 (16)	C21—C20—C25—C24	1.8 (3)
C18—C7—C8—C15	1.8 (3)	C19—C20—C25—C24	-178.52 (18)
C7—C8—C9—C14	0.51 (19)	C8—C7—N1—C14	-3.13 (18)
C15—C8—C9—C14	177.28 (15)	C18—C7—N1—C14	179.82 (14)
C7—C8—C9—C10	-179.41 (18)	C8—C7—N1—S1	-169.56 (12)
C15—C8—C9—C10	-2.6 (3)	C18—C7—N1—S1	13.4 (2)
C14—C9—C10—C11	1.0 (3)	C13—C14—N1—C7	-177.09 (18)
C8—C9—C10—C11	-179.08 (19)	C9—C14—N1—C7	3.40 (18)
C9—C10—C11—C12	1.0 (3)	C13—C14—N1—S1	-10.3 (3)
C10—C11—C12—C13	-2.2 (4)	C9—C14—N1—S1	170.19 (12)
C11—C12—C13—C14	1.2 (3)	O3—C15—O4—C16	1.7 (3)
C12—C13—C14—C9	0.9 (3)	C8—C15—O4—C16	-178.92 (16)
C12—C13—C14—N1	-178.54 (18)	C17—C16—O4—C15	-177.89 (18)
C10—C9—C14—C13	-2.0 (3)	C7—N1—S1—O2	-32.90 (16)
C8—C9—C14—C13	178.05 (17)	C14—N1—S1—O2	162.70 (14)
C10—C9—C14—N1	177.54 (15)	C7—N1—S1—O1	-162.21 (14)
C8—C9—C14—N1	-2.39 (18)	C14—N1—S1—O1	33.38 (16)
C7—C8—C15—O3	169.53 (19)	C7—N1—S1—C1	83.16 (15)
C9—C8—C15—O3	-6.5 (3)	C14—N1—S1—C1	-81.24 (15)
C7—C8—C15—O4	-9.8 (3)	C6—C1—S1—O2	17.63 (18)
C9—C8—C15—O4	174.12 (15)	C2—C1—S1—O2	-161.76 (15)
C8—C7—C18—C19	67.2 (3)	C6—C1—S1—O1	150.76 (16)
N1—C7—C18—C19	-116.4 (2)	C2—C1—S1—O1	-28.63 (18)
C7—C18—C19—C20	-176.79 (17)	C6—C1—S1—N1	-96.93 (16)
C18—C19—C20—C25	34.3 (3)	C2—C1—S1—N1	83.68 (16)

Hydrogen-bond geometry ( $\text{\AA}$ ,  $^\circ$ )

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C10—H10 $\cdots$ O3	0.93	2.48	3.000 (3)	116
C13—H13 $\cdots$ O1	0.93	2.33	2.908 (3)	120
C16—H16A $\cdots$ O1 <sup>i</sup>	0.97	2.56	3.381 (3)	143

Symmetry codes: (i)  $x-1, y, z$ .

Fig. 1

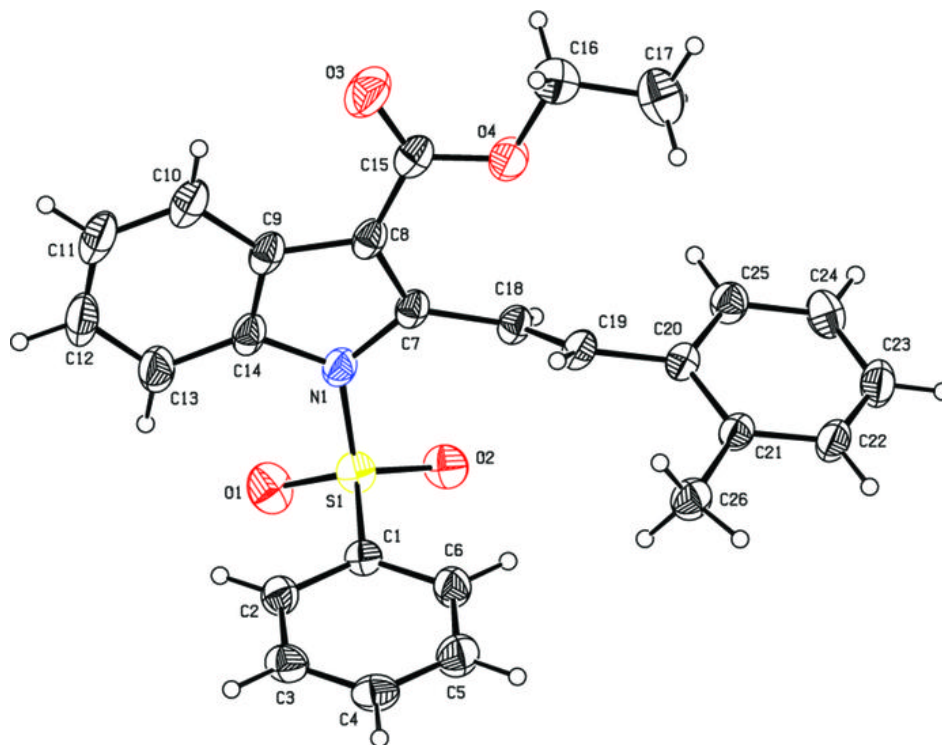


Fig. 2

